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Automatic Screening and Optimization of Electrochemical SPR (EC-SPR) based Graphane Batteries

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Abstract—The project is all about the concept of introducing graphane batteries in SPR based Plasmon technology. The main focus in my work is graphane batteries with the concept of electrochemistry. The main concept of EC is its indication for the rate of reaction is very high. By making use of two software battery design studio my work is focused in charging and discharging of the battery and giving high accuracy by making it cost effective. In the battery design studio by solving the drude model of the electrochemical concept and replacing the anode and the cathode layer by graphane solvent. The conductivity of graphane is found to be more efficient than the lithium batteries. The disadvantage of the lithium is due to SEI (solid electrolyte interface) which is the main reason for the heating up of the battery. The implementation with the solvent being graphane is implemented in the battery design studio and it is observed that the battery is charging for up-to 25 min duration but the discharging is taking up-to 75-100 hrs for the 20000 mah design batteries. The disadvantage is that the discharging cycle is faster when the temperature is above 230C. The complication of the batteries is increasing drastically.

Keywords— electrochemistry ,graphane , drude model,chemical model ,SEI

I. INTRODUCTION

Electrochemical systems have a quantitative formalism developed for the study of electrochemical. The function convolution in current from the electrochemical was found to be having the signal present in the chemicals therefore, in techniques used for electrochemical based current provides information and tools like EC-SPR is powerful. We can say that the concept of electrochemical is very co-operative to the physical and environmental condition such as temperature, pressure, vibration, power at different medium. Variations due to operating the chemical concept in terms of power is expensive to maintain the assumptions. Power management techniques as per the module voltage demands the ability to cross check the charging and discharging cycle including the solid electrolyte interface (SEI) implementation. Recent days the enhancement of the battery is enlarging very rapidly, in the research level also the analogy is still into considerations. From the late 18th century the daniel cell of the batteries is made the VLSI design industry have enhanced this chemical process in a very wide way. According to the history of the batteries the surge of interest on plasmonic batteries was not intended. Now the nm technology in decreasing the batteries size as well as reduce the charging cycle and increase the discharging cycle.

In general the concept of electrochemical have three different layers the anode cathode and separator three of this have their own specification the anode is the negative conductivity and cathode is the positive conductivity the glass is the normal silicate material as the Si metal is a transparent. The negative charge easily accumulates in the discharging layer that is the anode and slowly discharges this entire procedure of the electrochemical concept. It consist of an heterogeneous reactions, reacting between two chemicals. The solid electrolyte interface (SEI) formation inside the process is the main reason for effecting the energy density and also for the interaction of the fluids inside. From the chemical reactions that has been occurred they get converted to electrical energy. External voltage is responsible for the charging process.

II. DESIGN AND OPERATION PRINCIPLE

In the design the the anode and cathode will be of a graphene layer because graphene is found to be light, durable and high efficient in storing the data also extends the battery life lion is enhanced in the layers in graphene the capacity is found to be in 28 giga watt/hour the charging and discharging cycle is changed the charging is 200 cycles and discharging 3200 cycles this will increase the charging speed and also give longest durability the three layer in this design is known to be power layer, energy With the help of the software the design will be based on the 20000 mah . there are four different types of batteries but electrochemical is found to be more effective for the design. The design for 1500mah is done layer by layer the lion battery previously used has three different layer the anode separator and cathode the lithium phosphate is used in the anode and cathode layer the separator is made up of glass

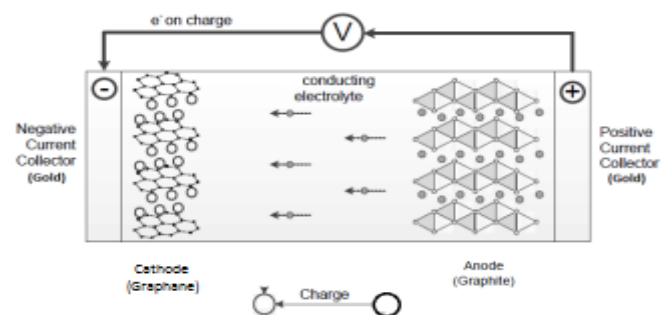


Fig. 1. Deposition of the anode and cathode level.

The design shows the internal chemical reaction to be occurred where the anode and cathode level depicts the positive and negative region respectively.

III. PURPOSED STRUCTURE

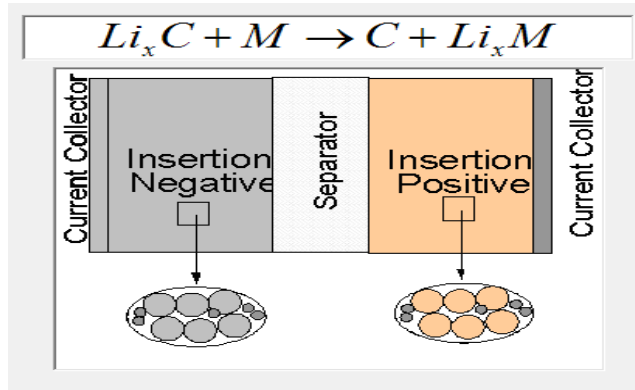


Fig. 2. Top view of the electrochemical-spr

SPR design for different layer in order to reduce the compact design to a thin film the layer by layer chemicals is placed in order to show the storing of charge in the layer deposition of the anode and discharging at the cathode hence when the SPR design is made the battery design reduces and the comactivity increase. In terms of the VLSI design ,it can be implemented in Lab-On-Chip. The main use of using SPR technology is once the VLSI coming to an end after 5nm technology the plasmon is coming to existence hence forth it can be implemented.

IV. SIMULATION RESULTS

The simulation results obtained front the BDS tool to detect the nonplasmonic range of the graphane batteries in order to store more of the data with respect to charging and discharging.

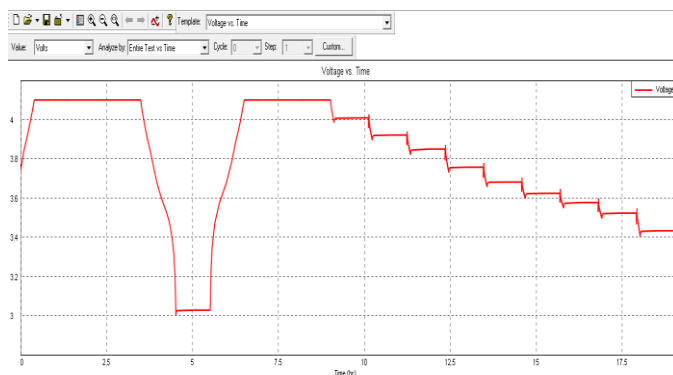


Fig. 3. Voltage vs time .

The above figure shows that the drop of the charging with respect to the discharging cycle. The time with respect to the 18 (hr) the charging and discharging of the batteries showing the dip at the 5 (hr) time period the voltage operating at the 2V

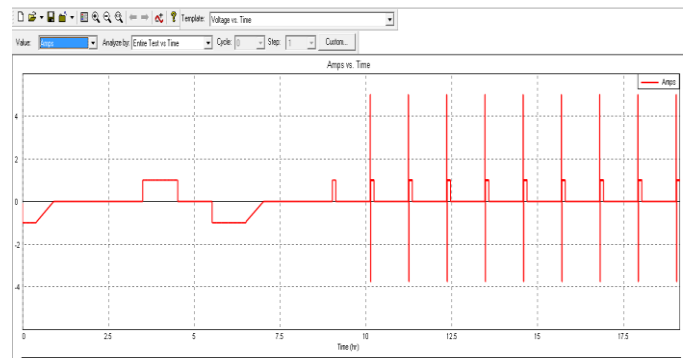


Fig. 4. Voltage vs time with amplitude

For the above graph the amplitude is into the consideration during the 18 hours of the discharging process the initial 10 hours of discharge cycle is very less but after the 10 hours due to the heating capacity of the battery the cycle toggles from the graph as shown. From the toggling cycle it is clear that the temperature plays a vital role.

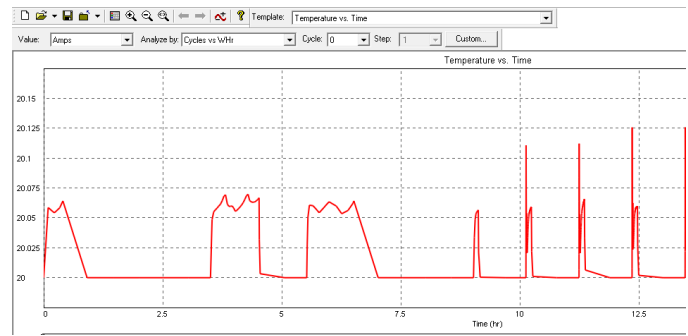


Fig. 5. Temperature vs Time

It is found that from the graph due to the temperature of the phone the charging and the discharging keeps varying a lot in the x-axis when the temperature exceeds 20⁰ C at room temperature. In the y-axis the time (hr) is up-to 12hrs only when the temperature of the battery is very high.

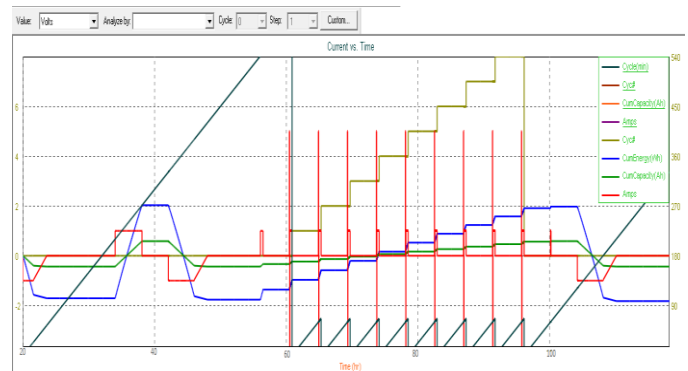


Fig. 6. Overview graph of all the parameters .

The initial discharging from the 0 to 10 hrs the charging vs discharging graph of all the parameters is to be obtained the cycle is linear. The discharging from the 10 to 20 hrs of duration the cycle is faster hence it is found that the charging cycle to be obtained is very fast.

Confidence Intervals:

90% = -0.246728 to 0.246728

95% = -0.293995 to 0.293995

99% = -0.386374 to 0.386374

EC-SPR revealed the kinetic parameters of the dissociation rate, SEI formation rate, mass, and thickness of all the layers. This in situ EC-SPR can display all the reactions on the surface of an electrode and we anticipate that this technique can be used to design appropriate SEI and electrolyte systems that are compatible with specific electrode materials.

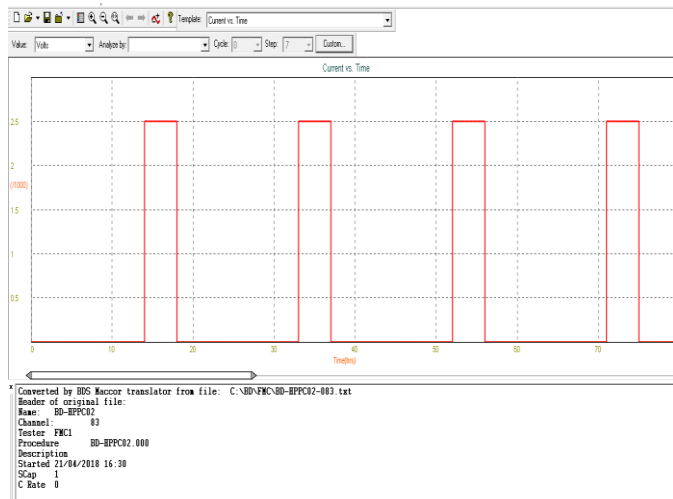


Fig. 7. Overview graph of all the parameters.

The initial discharging from the 0 to 10 hrs the charging vs discharging graph of all the parameters is to be obtained the cycle is linear. The discharging from the 50 to 80 hrs of duration the cycle is faster hence it is found that the charging cycle to be obtained is very fast.

V. CONCLUSION

The plasmonic based wave guide sensors were designed and are tested for the detection of battery that has been

designed in the plasmons as thinking into consideration it is known that the battery is facing a lot of trouble in the environment. The plasmonic batteries are designed in the SPR technology where the light is working as the source which is said to be travelling in the speed of 3×10^{-8} . The spr design of the battery is made in the R-Soft CAD suite. These batteries will reduce the heating power and also being cost effective. The durability will be very high and it can be implemented in any of the design the fabrication to be done by defect engineering using Lab-on chip.

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REFERANCE

- [1] Warwick Manufacturing Group, International Digital Laboratory, University of Warwick, "Electrochemical-Thermal Modelling and Optimisation of Lithium-Ion Battery Design Parameters Using Analysis of Variance". unpublished
- [2] Batteries and Electrochemical Capacitors by Daniel A. Scherson and Attila Palencsár, Nansi Xue "Design and Optimization of Lithium-Ion Batteries for Electric-Vehicle Applications" unpublished
- [3] Sylvia Ayu Pradanawati, Fu-Ming, Wang Chia-Hung Su, "Using electrochemical surface plasmon resonance for in-situ kinetic investigations of solid electrolyte interphase formation in lithium ion battery" vol. A247, pp. 529-551, April 2010.
- [4] C.K. Kim, D.S. Shin, K.E. Kim, K. Shin, J.J. Woo, S. Kim, S.Y. Hong, N.S. Choi, "Fluorinated hyperbranched cyclotriphosphazene simultaneously enhances the safety and electrochemical performance of high-voltage lithium-ion batteries", Chem Electro Chem (2016).
- [5] Robert Spotnitz Battery Consultant " Understanding Impedance of Li-Ion Batteries with Battery Design Studio", IEEE Transl. J. Magn. Japan, vol. 2, pp. 740-741, August 2015.
- [6] Richard Johns, "Modeling of Battery Systems and Installations for Automotive Applications", Oxford: Clarendon, 1892, pp.68-73.
- [7] Robert Spotnitz "Thermal & Electrochemical Simulation of Battery Pack Systems Steve Hartridge", CA: University Science, 2006.
- [8] Robert J., "Multilayer Electrode Coatings" Star Korea Conference, 2015