

# Optimized Hybrid Electrochemical and Magnetic Abrasive Finishing of Biomedical Alloys: A Critical Review of Process Parameters, Performance, and Future Directions

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**Abstract—** The demand for ultra-smooth, defect-free surfaces on metallic biomedical implants has exposed the limitations of conventional polishing techniques when applied to difficult-to-machine alloys such as Ti-6Al-4V, 316L stainless steel, Co-Cr-Mo, and magnesium alloys. Hybridization of Electrochemical Machining (ECM) with Magnetic Abrasive Finishing (MAF) combines contact-free anodic dissolution with flexible magnetic-brush polishing, offering simultaneous high material removal rate (MRR) and nanoscale surface finish without thermal damage or residual stress. This paper critically reviews process parameters (electrolyte chemistry, current density, magnetic flux density, abrasive characteristics, gap, and auxiliary energy fields), their influence on surface integrity, MRR, and biocompatibility of implant-grade alloys. Recent advances in eco-friendly electrolytes, real-time monitoring, and intelligent control are highlighted. Although surface roughness values as low as 30–50 nm Ra have been consistently achieved, challenges remain in scaling the process to complex 3D geometries and porous additive-manufactured implants. Directions for future research toward closed-loop, sustainable, and patient-specific finishing platforms are proposed.

**Keywords—** ECM-MAF hybrid finishing, biomedical implants, Ti-6Al-4V, 316L stainless steel, surface roughness, biocompatibility, green electrolytes

## I INTRODUCTION

Surface topography of metallic implants directly governs osseointegration, wear of articulating components, and long-term corrosion behavior in vivo [1,2]. Roughness exceeding ~100 nm Ra has been correlated with delayed bone apposition and increased inflammatory response [3]. Titanium alloys, medical-grade stainless steels, cobalt-chromium, and biodegradable magnesium therefore require finishing processes capable of delivering mirror-like surfaces while preserving bulk mechanical properties.

Electrochemical machining (ECM) removes material through controlled anodic dissolution without generating heat-affected zones or residual tensile stress [4]. Magnetic abrasive finishing (MAF) employs magnetically energized abrasive particles that act as a flexible brushing tool, achieving nanoscale smoothness even on freeform surfaces [5]. Sequential or simultaneous integration of ECM and MAF (commonly termed ECM-MAF or UECM-MAF when ultrasonic vibration is added) exploits the strengths of both processes: ECM rapidly removes macro-scale stock and passivates the surface, while MAF eliminates oxide layers and micro-protrusions left after electrolysis [6–8]. This review consolidates the state-of-the-art in ECM-MAF applied to biomedical alloys, identifies optimum parametric windows, evaluates sustainability aspects, and outlines research gaps that must be addressed for clinical translation.

## II PRINCIPLE AND CONFIGURATIONS OF ECM-MAF HYBRID PROCESS

In typical setups, the workpiece acts as anode and a cathodic tool is placed at 0.2–1 mm inter-electrode gap. Electrolyte (usually NaNO<sub>3</sub> or NaCl-based—is pumped through the gap while DC or pulsed current is applied. Flexible magnetic abrasive particles (MAPs—usually iron + Al<sub>2</sub>O<sub>3</sub> or SiC) are introduced into the working zone and pressed against the anode surface by an external magnetic field (0.3–1.0 T). Material removal occurs by combined anodic dissolution and micro-cutting/micro-ploughing by MAPs.

Two main configurations exist:

- I. Sequential ECM followed by MAF (most common)
- II. Simultaneous ECM-MAF (higher productivity, more complex electrolyte management)

Ultrasonic Vibration (20-40 KHz) is frequently superimposed to improve electrolyte renewal and abrasive

particle activation, leading to 40-80% improvement in final  $R_a$  [9, 10].

### III Influence of Key Process Parameters on Biomedical Alloys

**Table 1: Typical parametric ranges and best-reported performance of ECM–MAF on biomedical alloys.**

| Alloy          | Electrolyte                      | Current density (A/cm <sup>2</sup> ) | Magnetic field (T) | Abrasive & size                             | Auxiliary energy        | Initial $R_a$ (μm) | Final $R_a$ (nm) | MRR (mm <sup>3</sup> /min) | Reference(s) |
|----------------|----------------------------------|--------------------------------------|--------------------|---------------------------------------------|-------------------------|--------------------|------------------|----------------------------|--------------|
| Ti–6Al–4V      | 15 % NaNO <sub>3</sub>           | 10–20                                | 0.5–0.7            | Al <sub>2</sub> O <sub>3</sub> –Fe, 3–5 μm  | 20 kHz ultrasonic       | 0.8–1.2            | 30–80            | 0.15–0.45                  | [14,19,26]   |
| 316L SS        | 10–20 % NaNO <sub>3</sub> + NaCl | 8–15                                 | 0.4–0.6            | Diamond–Fe, 1–3 μm                          | Ultrasonic + rotation   | 0.6–1.0            | 35–50            | 0.20–0.38                  | [9,20,27]    |
| Co–Cr–Mo       | 15 % NaNO <sub>3</sub>           | 12–18                                | 0.5–0.8            | SiC–Fe, 5 μm                                | None or rotation        | 0.7–1.1            | 50–70            | 0.18–0.30                  | [21,28]      |
| Mg (AZ31/AZ91) | 10 % NaNO <sub>3</sub> + citrate | 5–10                                 | 0.3–0.5            | Al <sub>2</sub> O <sub>3</sub> –Fe, 5–10 μm | Low-frequency vibration | 1.0–1.5            | 90–120           | 0.08–0.15                  | [22,29]      |

**A. Electrolyte composition and concentration** Neutral salts (NaNO<sub>3</sub>, NaCl) at 10–20 wt% are preferred over acidic electrolytes to avoid pitting and hydrogen embrittlement in titanium and magnesium [11,12]. Recent studies show that mixed NaNO<sub>3</sub> + NaCl electrolytes with organic additives (glycerol, citrate) reduce toxic sludge by >60 % while maintaining uniform dissolution [13].

**B. Current density and voltage** 5–20 A/cm<sup>2</sup> (or 20–60 V pulsed) provides the best balance between MRR and surface quality for Ti–6Al–4V and 316L [14,15]. Excessively high current density (>25 A/cm<sup>2</sup>) causes sparking and localized boiling, deteriorating finish.

**C. Magnetic flux density and pole design** 0.4–0.7 T is optimum. Higher fields increase normal force on MAPs and risk abrasive embedding in soft Mg alloys [16].

**D. Abrasive type, size, and proportion** Diamond or Al<sub>2</sub>O<sub>3</sub> particles (1–10 μm) bonded with carbonyl iron (50–150 μm) in 20–40 vol% ratio yield best results. Unbonded MAPs give higher MRR but poorer finish [17].

**E. Auxiliary energy (ultrasonic, rotation, vibration)** 40–60 % roughness reduction and 2–3× MRR enhancement reported with 20 kHz ultrasonication [9,18].

#### IV Performance Achieved on Major Biomedical Alloys

Ti–6Al–4V shows a remarkable drop in surface roughness from an initial  $R_a$  of 0.8–1.2 μm to a final 30–80 nm, with material removal rates of 0.15–0.45 mm<sup>3</sup>/min and no formation of α-case or β-phase depletion [8,14,19]. On 316L stainless steel, the process improves surface roughness from ~600 nm to 35–50 nm in only 12–20 minutes, simultaneously reducing corrosion current density by 65–80 % in simulated body fluid [9,20]. For Co–Cr–Mo alloy, MAF achieves  $R_a$  below 60 nm in a single simultaneous operation, resulting in approximately 40 % lower wear rate against UHMWPE [21]. Magnesium alloys (AZ31/AZ91) also benefit from a combined

controlled-dissolution and immediate MAF polishing approach that prevents excessive corrosion and pitting during processing, yielding a final  $R_a$  around 100 nm with a clean surface [22].

#### V Sustainability and Green Approaches

Conventional electrochemical machining (ECM) typically produces substantial amounts of metal-hydroxide sludge. However, recent advancements have significantly addressed this issue: Electrolyte recirculation combined with membrane filtration systems can reduce waste generation by more than 70 % [23]. The use of biodegradable citrate-based electrolytes offers an environmentally friendlier alternative [13]. Additionally, integrating cryogenic assistance with MAF markedly lowers electrolyte consumption during the process [24].

#### VI Challenges and Future Research Directions

Despite promising results, several gaps remain:

1. Lack of robust real-time monitoring (current, temperature, pH, conductivity) and closed-loop control.
2. Limited data on finishing of porous/lattice structures produced by metal additive manufacturing.
3. Absence of standardized in-vitro/in-vivo studies linking specific ECM–MAF surface textures to cellular response.
4. Scale-up to industrial batch production while maintaining consistency.

Integration of machine learning for parameter optimization [25] and digital-twin-based process simulation offers immediate pathways forward.

#### VII CONCLUSION

Hybrid ECM–MAF has evolved into one of the most effective finishing techniques for biomedical alloys, routinely

delivering sub-50 nm surfaces without inducing thermal or mechanical damage. With continued focus on green electrolytes, sensor integration, and intelligent control, the process is poised for widespread adoption in the manufacture of next-generation orthopedic, dental, and cardiovascular implants.

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